## SEPARATION OF A MIXTURE OF *R*-MENTH-4-EN-3-ONE AND (-)-MENTHONE

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A promising chiral synthon is R-menth-4-en-3-one (1) [1-4], which is produced from the natural monoterpenoid l-menthol through the intermediate (-)-menthone (2) [5, 6] and is often contaminated (up to 10%) with the latter. The usual purification methods, including chromatographic ones, do not produce 1 of the desired purity.

It is known [7] that  $\bf 2$  does not form bisulfites. Therefore, we supposed that it might be possible to isolate  $\bf 1$  from a mixture with  $\bf 2$  by forming the bisulfite derivative (3) of the carbonyl. According to our previous research [1], nucleophilic addition to  $\bf 1$  should occur via 1,2-addition to the carbonyl. Furthermore, the reaction was performed under the literature conditions [8, 9] in order to prevent 1,4-addition of NaHSO<sub>3</sub>.

In fact, treatment of a mixture (9:1) of **1** and **2** with a four-fold excess of NaHSO<sub>3</sub> followed by standard workup gave chemically pure (>99%) **1** in 92% yield. The 1,4-addition product (**5**) was not observed in the reaction mixture.

Thus, pure 1 could be isolated from its mixture with 2 by using the unusual behavior of 1 in nucleophilic addition reactions.

**Separation of R-Menth-4-en-3-one from (-)-menthone.** A mixture  $(9.20\,\mathrm{g}, 9:1)$  of 1 and 2 was treated with a solution of Na<sub>2</sub>SO<sub>3</sub>  $(27.34\,\mathrm{g}, 217\,\mathrm{mmol})$  in H<sub>2</sub>O  $(50\,\mathrm{mL})$  and several drops of phenolphthalein. The reaction mixture was stirred at 95-97°C for 1 h adding dropwise AcOH  $(25\,\mathrm{mL}, 50\%)$  until the color faded and was extracted with petroleum ether  $(3\times30\,\mathrm{mL})$ . The extract was dried over Na<sub>2</sub>SO<sub>4</sub> and filtered. The solvent was evaporated in vacuo. Compound 5 was not observed in the solid. The aqueous layer was treated with NaOH solution (30%) and extracted with methyl-*t*-butylether  $(3\times30\,\mathrm{mL})$ . The extract was dried over Na<sub>2</sub>SO<sub>4</sub> and filtered. The solvent was evaporated in vacuo to afford 2  $(7.64\,\mathrm{g}, 92\%)$  that was pure according to GC, PMR, and <sup>13</sup>C NMR [10].

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Chromatographic analysis was performed on a Chrom-5 instrument [column 2.4 m, stationary phase PEG-6000 (5%) on Inerton AW-DMSC (0.125-0.160 mm), working temperature 50-200°C] using He carrier gas. NMR spectra ( $\delta$ , ppm, J/Hz) were obtained on a Bruker AM-300 spectrometer (working frequency 300.13 MHz for PMR and 75.47 MHz for  $^{13}$ C) in CDCl<sub>3</sub> relative to TMS.

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